

Update on Improvements to Dissolved Hydrocarbon Gases in Water Analysis

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Why measure hydrocarbon gases in groundwater?

- Oil & gas application
 - Predrill site characterization
 - Establish baseline for comparison
 - Find existing issues
 - Post drill site characterization
 - Assess potential changes
- Remediation monitoring
 - Reducing conditions indicator
 - Active reductive dechlorination
 - Monitored natural attenuation
- Other applications?





How to measure hydrocarbon gases in groundwater?

- Headspace GC / FID
 - ~ RSK-175 Rev 0, 1994
 - J Chrom Sci Kampbell, Vandergrift, 1998
 - ~ M E E Analysis Guidance, US EPA Region 1, 2002
 - ~ RSK-175 Rev 2, 2004
 - ~ RSK-175 Rev 3, 2006
 - ~ RSK-175 Rev 5, 2010
 - ~ PA DEP 3686 Rev 1, 2012
- Purge and Trap
 - ~ PA DEP 9243 Rev 0, 2012

RSKSOP-175 Revision No.2 May 2004 Page 1 of 14 Felisa Hudson

STANDARD OPERATING PROCEDURE

Sample Preparation and Calculations for Dissolved Gas Analysis in Water Samples Using a GC Headspace Equilibration Technique

Disclaimer

This standard operating procedure has been prepared for the use of the Ground Water and Ecosystems Restoration Division of the U.S. Environmental Protection Agency and may not be specifically applicable to the activities of other organizations. THIS IS NOT AN OFFICIAL EPA APPROVED METHOD. This document has not been through the Agency's peer review process or ORD clearance process.

Purpose (Scope and Application):

This method is applicable to the preparation of water samples for determination of dissolved gases. After quantitation of gas equilibrated into the prepared headspace, this method permits calculation of the concentration of the dissolved gas in the water before equilibration. Resulting concentrations are expressed as mg/L and µg/L of dissolved gas in water. This method has been used for determining dissolved hydrogen, methane, ethylene, ethane, propane, butane, acetylene, nitrous oxide and oxygen. The number of analyses that can be performed in an eight



- Lab to lab implementation variability of RSK-175 method
- No commercially available proficiency test samples
- Calibration standards are gas phase
 - Samples are water
 - Thus standards and samples are not handled identically
- Analyte loss during sample preparation
- Maintaining analyte representativeness during sampling



- Lab to lab implementation variability of RSK-175 method
 - Critical process elements
 - Sample transfer between containers
 - Headspace development
 - Equilibration
 - time
 - temperature
 - mixing
 - ~ Process options
 - ° Manual
 - Semi-automated
 - Fully automated





- No commercially available proficiency test samples
 - No catalog items
 - No custom PTs available either
 - Limited water solubility
 - Limited organic solvent solubility
 - Limited stability in water
 - Complex preparation process
 - = expensive to produce, short shelf life & lower quality
 - >> No independent check on accuracy



- Calibration standards are gas phase
 - Samples are water
 - Thus standards and samples are not handled identically
 - Equilibrium goes in one direction for samples and the opposite for standards
 - Non-equilibrium conditions
 - High biased standard response
 - Low biased sample response
 - = low biased calculated sample concentration





- Analyte bias during sample preparation
 - Pierced septum leaks
 - Headspace / analyte loss while waiting in autosampler





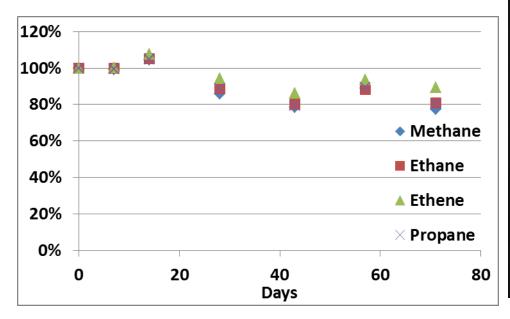
- Multi-lab validated method
 - ASTM D8028Final Ballot 2017
 - Single lab validation at present
 - Fully automated process with manual option
 - Marcellus Shale Coalition
 - Reviewed draft ASTM method
 - Identified inter-lab variability as a critical issue





Sample Storage

- Zero headspace vial
- Butyl rubber septum
- Sulfuric acid preservative
- 2 6 °C







- Water based calibration standards and QC samples
 - In-lab prep of single analyte saturated water standards
 - Use literature values at known temperature and pressure
 - PA DEP 3686 Rev 1, 2012
 - ASTM D8028
 - Immediate dilution to working standards
 - Store in VOC vials no headspace, 14 days
 - Commercial providers attempting to develop standards
 - LGC is working on reference materials
 - Water CRM for MEEP compounds
 - Organic solvent CRM for
 - acetylene, n-butane, isobutylene, 2-methylpropane



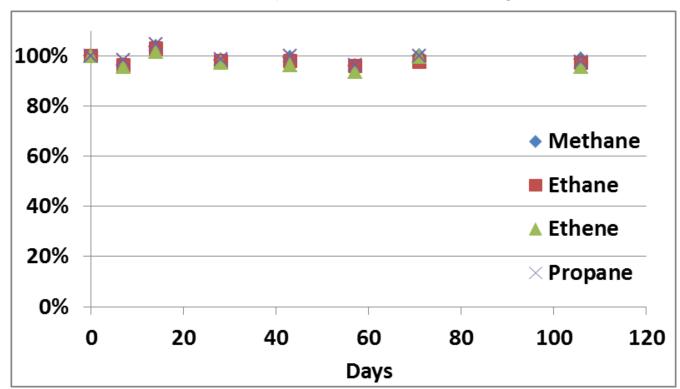
Methane 23.2 mg/L

Ethane 62.0 mg/L

Ethene 149.0 mg/L

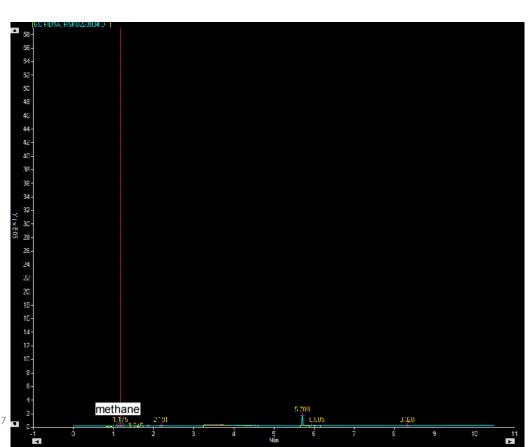


- Water based calibration standards and QC samples
 - TestAmerica
 - 4 analyte formulation @10 mg/L
 - Months of stability alternative storage container



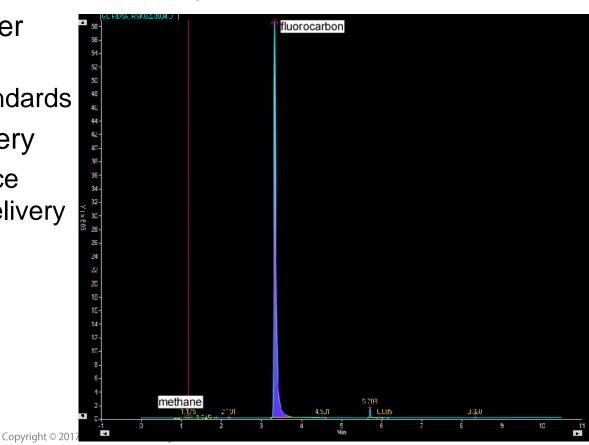


- Surrogate monitor sample prep & analysis quality
 - How to know when you have a good sample prep and headspace injection?
 - ~ Int. Std.
 - Surrogate
 - ~ ASTM
 - MTBE
 - ~ HFCs



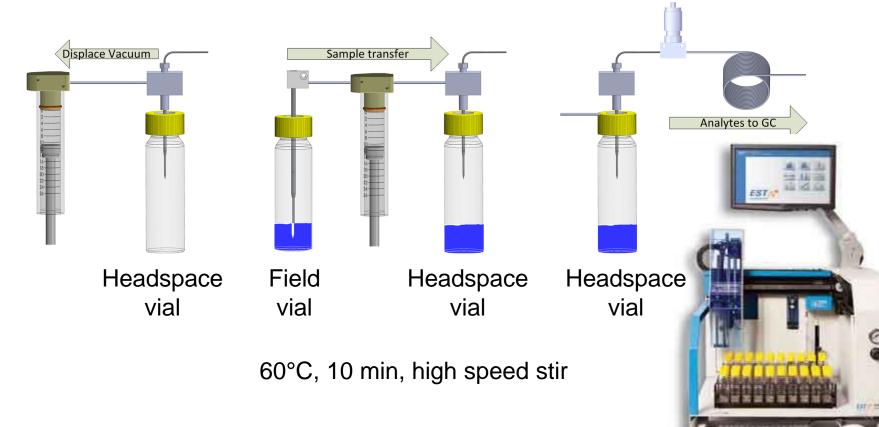


- Surrogate monitor sample prep & analysis quality
 - How to know when you have a good headspace injection?
 - Surrogate in water
 - Prepare like calibration standards
 - Automated delivery
 - Zero headspace storage and delivery



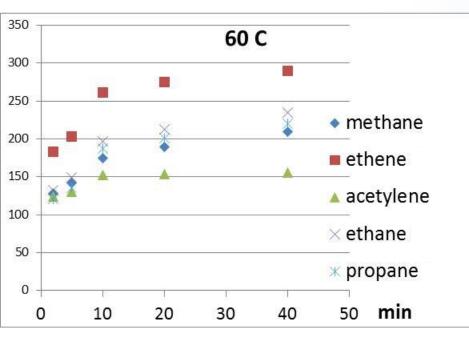


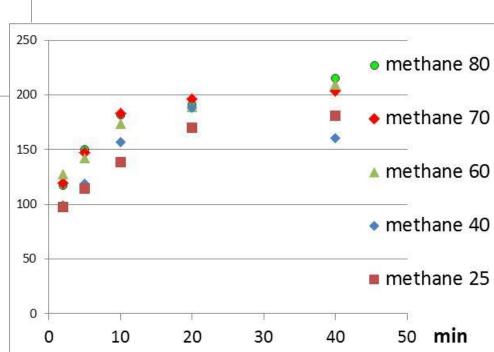
- Automated headspace preparation
 - EST Analytical LGX 50





Headspace Equilibrium Optimization







Selecting a Stir Bar

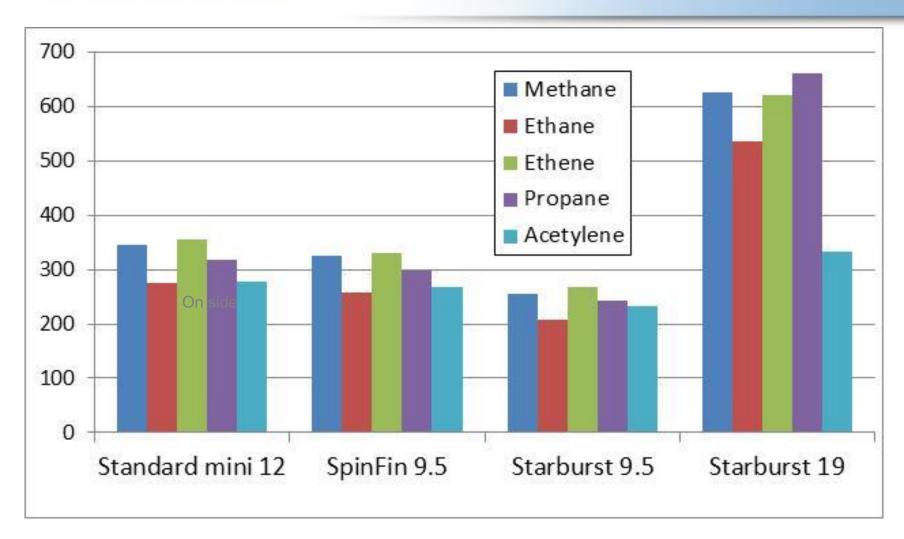




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Selecting a Stir Bar



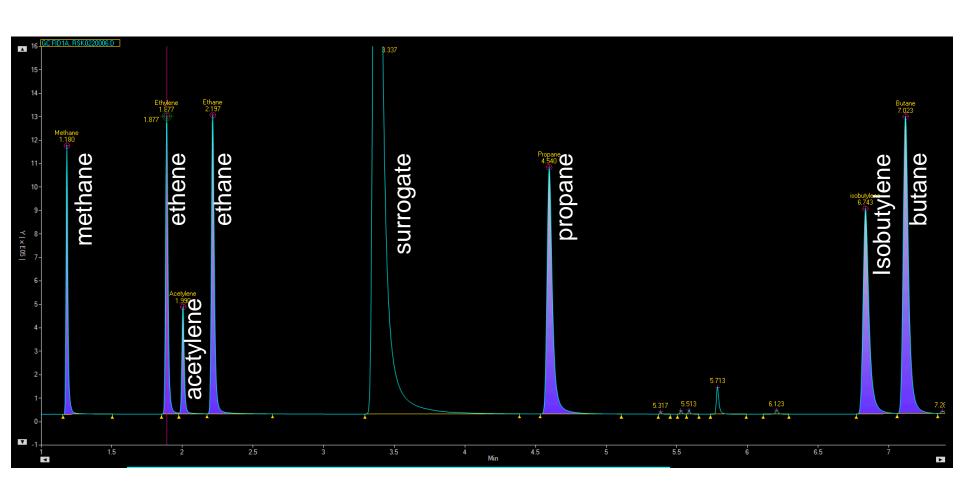


Selecting a GC Column

- Consider effects of water and carbon dioxide
 - Retention time shifts
 - Acetylene
 - Ethylene
- Salt columns water content of phase effects RT
- PLOT columns less water retention better stability



Stable Separation (Q PLOT)





External Proficiency Test samples

Future state –
Regular
independent
accuracy
verification



Current state No
independent
accuracy
verification



Summary

- Multi-lab validated method
 - On track ASTM D8028
- Water based calibration standards and QC samples
 - Open opportunity for commercial provider (LGC)
 - Lab based saturated stock water solutions
- Surrogate monitor sample prep & analysis quality
- Automated headspace preparation
 - ~ 60°C, 10 min, high speed stir with 19 mm starburst
- External Proficiency Test samples
 - Open opportunity for commercial provider (LGC)



Acknowledgements

TestAmerica

Mike Dunn
Brandon Matthews
Ray Risden
Matt Smith
Daniel Waite



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